### HETEROGENEITIES AT THE CELL LEVEL BY MODELING AND EXPERIMENTS



Dietmar Gerteisen, Nada Zamel Fraunhofer Institute for Solar

Energy Systems ISE

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### AGENDA

- Fuel Cell Systems Department at Fraunhofer ISE
- Motivation
- Spatially Resolved Analysis
  - The Multi-Channel Characterization System (MCCS)
  - The Segmented Cell
- Local Measurement of Current, Voltage, EIS and HFR
- Special Cases
  - Local Perturbation
  - Hydrogen Evolution on the Cathode
- Conclusion / Outlook



# The Fraunhofer-Gesellschaft is the largest organization for applied research in Europe.

- 22,000 employees
- 66 institutes & research units
- Budget 2 billion €
- >70% with contract research
- growing international activities





### Our department Fuel Cell Systems is assisting industry for over 20 years now.

- Budget 2.2 Mio. € (2013)
- 7 engineers, 5 scientists, 2 PhD students, 1 Technician, ca. 15 students
- Markets:
  - automotive fuel cells
  - back-up power
  - portable generators & micro fuel cells





### References





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# Technological breakthroughs need a new understanding of processes with high time and spatial resolution.

- Fuel cell electrical vehicle with long range
- Performance specifications and cost targets are hard to reach
  - Power/cell:
    > 1 W/cm<sup>2</sup> @ 675mV
  - Total operation time: > 5000 h
  - Freeze start capability:
    @ -20 °C
  - Cost per kW:
    30-40 €



Test drive with Daimler F-Cell of Fraunhofer ISE.



Different coupled processes within a fuel cell lead to inhomogeneities over the active cell area.



- Declining concentration of oxygen and hydrogen along the channel
- Rising humidity and temperature of gases along the channel
- Rising temperature of cooling liquid from inlet to outlet
- To optimize cell design, materials, and operation strategy these inhomogeneities must be understood



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# Schematic set-up of the MCCS with the segmented fuel cell

Every segment is loaded by its own potentiostat (synchronized)

Every potentiostat communicates with its own FRA





# 68-channel-characterization system for spatially resolved analysis of electrochemical energy converters

- Potentiostats:
  50 x (+/- 5 A; +/- 5 V) &
  18 x (+/- 30 A / +/- 5 V)
  together with 50 +18
  FRAs
- Frequencies: 0.1 Hz to 10 kHz
- current / voltage
  mapping
  (up to 790 A)
- el.-chem. impedance spectroscopy
- chronovoltammetry, chronoamperometry





### For spatially resolved cell characterization we segment the bipolar plate.

- Electrical isolation of the bipolar plate
- Electrical isolation of GDL and electrode possible
- Up to 68 segments



segmentation of cells



contact pins



# Electrochemical impedance spectroscopy (EIS) is measured at a specified steady state point.





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- Active area of 49 cm<sup>2</sup> (7 x 7 square matrix)
- Insulation between the segments achieved with epoxy
- 3 channel serpentine flow field
- Gore membrane with a SIGRACET GDL 25 BC



# The hydration level of the cell can be monitored via its spatially resolved OCV humidity and flow rate affect hydration of the membrane





Gerteisen et al. Int. J. Hydrogen Energy, 37 (2012) 7736-7744

The hydration level of the cell can be monitored via its spatially resolved HFR and current density flow configuration and its effect on hydration





# Spatially resolved characterization development of cell design and operation strategy

Current density is limited in the regions of the air channel bends

Performance is improved by a flow field redesign





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### Local Analysis of the electrochemical impedance spectra

**Operating conditions** 

- Anode: H<sub>2</sub> with  $\lambda_{H_2} >> 1$
- Cathode: air with  $\lambda_{O2}$  < 2 @ U<0.6V
- Co-flow mode
- Atmospheric pressure







### First explanation was given by Schneider et al.<sup>(1)</sup>

- 1. I.A. Schneider, S. A. Freunberger, D. Kramer, A. Wokaun & G.G. Scherer, J. Electrochem. Soc. 2007 154(4): B383-B388
- 2. T. Jacobsen, P. V. Hendriksen, S. Koch, Electrochimica Acta 2008, 53(25): 7500–7508
- 3. A. A. Kulikovsky, J. Electrochem. Soc. 2012 159(7): F294-F300

### Basic idea:

d.c. current per segment:  $i_{seg} = i_0 c exp(\eta/b)$ 

a.c. current response:  $\Delta i_{seg} = i_0 \exp(\eta/b) (\Delta c + c \Delta \eta/b) - C_{DL} d_t \Delta \eta$ 



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- MCCS has the capability to perturb specific segments
- Perturbation of the whole cell (all segments)





- MCCS has the capability to perturb specific segments
- Perturbation of col #1,2,3,6,7



- MCCS has the capability to perturb specific segments
- Perturbation of col #6,7





- MCCS has the capability to perturb specific segments
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Zamel et al. Fuel Cells, 13 (2013) 910-916

### EIS-Simulation by a simplified 2+1D model





### EIS-Simulation by a simplified 2+1D model





Fuel cell segments are coupled to nearest neighbors (x-y-direction) by

- Gas convection of air and vapor according to flow field pattern
- Cross current in GDL
- In-plane permeation of liquid water



### **Stoichiometric impact on the local EIS**

Comparison of full perturbation vs local perturbation @ low frequency (10mHz) shows...

- Iarge deviance at λ<sub>air</sub> < 3</p>
- reduced artifact at high λ<sub>air</sub>
  - error<5% @  $\lambda_{air}$  > 7 @ 650mV
  - error<5% @ λ<sub>air</sub> > 15 @ 800mV

→ High stoichiometry needed for analysing mass transport processes in the porous transport layers

 $\rightarrow$  Otherwise this effect has to be taken into account by modeling





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A bistable behavior in the performance of the cell is measured under the steady state condition An increase in current is measured despite depletion of oxygen  $\lambda_{o2} < 1$ 



Spatially resolved polarization curves with steady-state operation





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Spatially resolved polarization curves with steady-state operation



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### Under transient conditions, transport limitations can be minimized **Description of Experiment**







- Conditioning @ 0.2V: establishing saturation distribution
- Super-fast pol-curve (upwards)
- **Dwell time 5s at high voltage: equalization of c\_{02}** distribution in active area  $\rightarrow$  gas residence time ~230ms
- Super-fast pol-curve (downwards)



### Under transient conditions, transport limitations can be minimized Description of Experiment



Transient operation means:

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A bistable behavior in the performance of the cell is measured under transient operation The recovery occurs at a higher voltage



- Spatially resolved polarization curves with transient operation
  - Again: Bending back of the current curve
  - Again: Current recovery at relatively high positive voltage
  - Characteristics are the same for all segments (independent of region: inlet, middle, outlet)

INLET SEGMENTS



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  - Characteristics are the same for all segments (independent of region: inlet, middle, outlet)

#### **OUTLET SEGMENTS**



### Summary of the experimental findings what can we learn from the measured polarization curves?

Based on experimental measurements the following is true:

- Vertical branch in the i-U curve of the cell corresponds to the stoichiometric possible current from the oxygen fed to the cell
- Current density in the tail of the i-U curve lies a little bit above limiting current
- The following can then be concluded:
  - The NDR branch originates from a cathodic process
  - This tail originates from an unexpected second Faradaic process



# Explanation for the measured findings (steady-state) theory of HER on the cathode



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### **Explanation for the measured findings (steady-state)** theory of HER on the cathode



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### **Explanation for the measured findings theory of HER on the cathode**





### **Explanation for the measured findings theory of HER on the cathode**





# But, how can we explain the transient behaviour?



### **Explanation of the transients**

description of time-dependent cathode model



HER mechanism: Volmer – Heyrovsky  $Pt^* + H^+ + e^- \rightleftharpoons Pt - H_{upd}$   $Pt - H_{upd} + H^+ + e^- \rightleftharpoons Pt^* + H_2$ 

Measureable current under dynamic operation

$$i_{meas} = \frac{i_{ORR}}{i_{HER}} + i_{HER} + i_{DL-charg}$$

ORR described by Tafel approach

• 
$$i_{ORR}[t] = 4 * F * i_0 * c_{O2}[t] * Exp\left[\frac{\Phi_e[t] - \Phi_p[t] - VOC}{b}\right]$$



### **Explanation of the transients description of time-dependent cathode model**



HER mechanism: Volmer – Heyrovsky  $Pt^* + H^+ + e^- \rightleftharpoons Pt - H_{upd}$   $Pt - H_{upd} + H^+ + e^- \rightleftharpoons Pt^* + H_2$ 

Measureable current under dynamic operation

 $\bullet i_{meas} = i_{ORR} + i_{HER} + i_{DL-charg}$ 

HER described by Volmer-Heyrovsky steps

• 
$$i_V[t] = k * F * ((1 - \theta_H[t]) * Exp\left[-\frac{\Phi_e[t] - \Phi_p[t] - \Delta\Phi}{b2}\right] - \theta_H[t] * Exp\left[\frac{\Phi_e[t] - \Phi_p[t] - \Delta\Phi}{b2}\right])$$

with equilibrium potential  $\Delta \Phi = \Phi^0 + \frac{RT}{F} \ln(\frac{a_{H+}(1-\Theta)}{\theta})$ 

• 
$$i_H[t] = h * F * (\theta_H[t] * Exp\left[-\frac{\Phi_e[t] - \Phi_p[t] - \Delta \Phi}{b2}\right] - (1 - \theta_H[t] *) * c_{H2}[t] * Exp\left[\frac{\Phi_e[t] - \Phi_p[t] - \Delta \Phi}{b2}\right])$$



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### **Explanation of the transients description of time-dependent cathode model**



- HER mechanism: Volmer Heyrovsky  $Pt^* + H^+ + e^- \rightleftharpoons Pt H_{upd}$   $Pt H_{upd} + H^+ + e^- \rightleftharpoons Pt^* + H_2$
- Measureable current under dynamic operation
  - $\bullet i_{meas} = i_{ORR} + i_{HER} + \frac{i_{DL-charg}}{i_{DL-charg}}$
- Charging of the double layer capacitance
- $i_{DL-charg}[t] = C_{DL} * \partial_t (\Phi_e[t] \Phi_p[t])$



### **Explanation of the transients Description of time-dependent cathode model**



Differential equation system with the solving variables  $\Phi_{p}$ ,  $C_{O2}$ ,  $\Theta_{H}$ 

Assuming no anode polarization  $\rightarrow \Phi_p$  @ anode = 0V

$$\Phi_p[t] = -R_{Ohm} * \left( i_{ORR}[t] + i_{HER}[t] + i_{DL-charg}[t] \right)$$

Fickian Diffusion in GDL, no spatial resolution of the CCL

$$\varepsilon_{GDL} * \partial_t c_{O2}[t] = -\frac{i_{ORR}[t]}{4*F} + D_{GDL}^{eff} * \frac{c_{O2}^{inlet} - c_{O2}[t]}{L_{GDL}}$$

Hydrogen as solely adsorbate

$$\Gamma * \partial_t \Theta_H[t] = i_V[t] - i_H[t]$$





















Operation

- Counter-flow
- Low to high voltage sweep
- Curve characteristics
  - More pronounced recovery branch
  - Negative currents
  - Influence of local position visible at entire pol-curve





### Can the model predict the trends?

The simulation results show the characteristics of back bending and recovery





### Can the model predict the trends?





# Impact of the model parameters analysis of the GDL porosity

Porosity determines the current overshoot due to the oxygen storage capacity





### Impact of the model parameters analysis of the ohmic resistance (mainly R<sub>mem</sub>)

- Ohmic resistance impacts the entire characteristic
- High resistance suppresses the HER → cathodic Galvani potential not low enough





### Impact of the model parameters

analysis of the reaction rate constant of the Volmer step

Reaction rate constant has a strong influence on the current recovery characteristics





# Impact of the model parameters analysis of the sweep duration

Considering the gas storage capacity, convergence to steadystate is reached @ ~ 2s





### Impact of the sweep direction

# model predicts the partly negative current @ upward sweep





Can the HER be used to explain the trends of the impedance spectra towards the outlet of the air stream? a second arc is measured despite local perturbation



- Operation
  - Counter-flow
  - Potentiostatic mode
  - Cell temperature kept at 60°C
  - i-U curve collected for a voltage range of 800 mV – 50 mV
  - 30 minutes at each voltage increment
  - Air/hydrogen flow rates 600/600
  - Inlet relative humidity of air/hydrogen 80%
  - Local perturbation of cell is used



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### Conclusion



- Due to the various and coupled processes, inhomogeneities are inevitable within the cell area
- Spatially resolved characterization is needed to disclose these inhomogeneities
- With the 68 channel system, characterization of various cell areas is possible
- Spatially resolved characterization allows for the investigation of specific trends
- Measurements with the MCCS can be coupled with our single cell EIS monitoring of short stacks, also under extreme climate conditions
- This helps to optimize design, material, and operation strategy



### Thank you for your attention!



Dr. Dietmar Gerteisen/ Dr. Nada Zamel Fraunhofer Institute for Solar Energy Systems ISE

dietmar.gerteisen@ise.fraunhofer.de / nada.zamel@ise.fraunhofer.de

www.h2-ise.com



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### Thank You Very Much for Your Attention!



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www.gecko-fuelcell.com



### Fraunhofer Institute for Solar Energy Systems ISE Dr. Dietmar Gerteisen

www.h2-ise.de

dietmar.gerteisen@ise.fraunhofer.de



### **Conclusion and Outlook**

- Theoretical explanation is given for the measured N-shaped polarization curves in steady-state as well as dynamic operation
- A simple 0-dim model that accounts for the oxygen storage capacity of the GDL, the ORR as well as HER on the cathode predicts the current back bending and recovery very well
- Fast transient measurements can be used to decompose mass transport losses from the residual losses (kinetics, proton migration, contact resistance,...) but
  - the sweep rates have to be adjusted to minimize in-plane effects (high sweep rates) and to minimize the HER on the cathode (low sweep rates)



### **Conclusion and Outlook**

- It has to be clarified if at low cell voltage the HER on the cathode affects the cathode impedance spectrum
- If hydrogen on the cathode side (generated at fast load steps or at undesired oxygen starvation operation) has an impact on cathode degradation has to be discussed



### Impact of the model parameters

# Assuming a linear drop of the oxygen concentration at channel | GDL boundary

Since gas residence time (230ms) is in the same range as the sweep time (120ms) downstream effects along the channel have to be considered







- PEMFC with 200 cm<sup>2</sup> active area
- Co-flow with meandric flowfield
- Characterization at 0.6 V, r.H. 50% (H<sub>2</sub>/air), T=60°C
- Variation of stoichiometry









Characterization at 0.5 V, r.H. 95% ( $H_2$ /air); stoichiometry of 2.2 /2.3 ( $H_2$ /air)



- Current mapping shows very low current production at inlet and outlet
- HFR indicates dehydration at the inlet
- Symmetric behaviour with respect of flow field design

