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Deformations caused by hydraulic pressure differences between flow battery half-cells

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BACKROUND & MOTIVATION

- Dynamic viscosity of anolyte and catholyte differs with temperature and state of charge (SOC) (shown in figure 1)
- Intensity of viscosity differences is depending on electrolyte composition (shown in table 1)
- Direct proportionality between viscosity and pressure losses lead to pressure differences between half-cells
- Deflections of flexible membranes und bipolar plates can be expected
- Deflections may vary over SOC and therefore cause plastic deformations of graphite felt



10 20 30 40 50 60 70 80 90 80 70 60 50 40 30 20 10 SOC [%]

Fig. 1 Varying dynamic viscosity of $V^{2+/3+}$ and $V^{4+/5+}$ electrolyte depending on SOC, 1.8 mol/L V, 2.7 mol/L H₂SO₄, 30 ℃ [1]

Composition	V ^{2+/3+}	V ^{4+/5+}
2 mol/L V 5 mol/L H ₂ SO ₄	3.80 mPa*s	2.90 mPa*s
1.6 mol/L V 2.6 mol/L H ₂ SO ₄	3.96 mPa*s	3.57 mPa*s

APPROACH

- Model design of the assembly: graphite felt – bipolar plate – graphite felt
- Deflections can be simulated using a modificated Kármán theory for nonlinear large bending of thin plates
- For determining the forces involved, the cell is separated in segments (shown in figure 2)
- Solving of the displacement function for finite matrix points
- Computation loop for the dependencies: felt compression – pressure loss
- Graphite felt electrodes can be assumed as nonlinear elastic-plastic spring assembly



Fig. 2 Flow direction and pressure segments of a model half-cell



electrodes

1.8 mol/L V 5.36 mPa*s 3.88 mPa*s 2.7 mol/L H_2SO_4

Important impact on the design of large-scale flow battery cells

Tab. 1 Dynamic viscosity of electrolytes at 30°C and 60% SOC [1;2]

- Establishing a theoretical computation basis
- Visualization of the shape and variance of thin plate deflections

RESULTS

SOC dependent deflections of the bipolar plate in the assembly graphite felt – bipolar plate – graphite felt for an all-vanadium flow battery cell with 135 cm² active area



Fig. 3 Deflections of the bipolar plate in the assembly graphite felt – bipolar plate – graphite felt, GFD 4.6 with 13% initial compression, 1.8 mol/L V, 2.7 mol/L H₂SO₄, 30 °C

CONCLUSION

- Pressure differences between flow battery half-cells cause deflections of flexible membranes und bipolar plates
- Deflections vary over the state of charge
- Contrary to the originally expected pear-shaped deformation (by the triangular load distribution), the actual deflection is more evenly distributed

OUTLOOK

- The simulation model will be adapted for the computation of larger flow battery cells, as much higher deflections are expected
- The deformation behavior in a battery stack will be investigated

DEDUCTIONS

SOC-dependent variance of membrane and bipolar plate deflections will intensify with

- increasing pressure losses by increasing volumetric flow, viscosity and flow length as well as decreasing flow area and felt permeability
- plastic deformations of graphite felt electrodes by fiber breakage

References

[1] Li, X.; et al.: Investigation of the use of electrolyte viscosity for online state-of-charge monitoring design in vanadium redox flow battery; Applied Energy, 2018 [2] Skyllas-Kazacos, M., et al.: Vanadium electrolyte studies for the vanadium redox battery - a review; ChemSusChem, 2016